Chapter II.9 Bacterial Growth and Antibiotics in Animal Respirometry

G.J. Dalla Via1

1 Introduction

The uncontrolled contribution of bacterial oxygen consumption in animal respirometry respresents a substantial problem and results in an ambiguous reading of the animal's metabolic rate. This problem became especially important when the application of polarographic oxygen sensors (POS) made possible long-term measurements of the dynamics of an animal's energy metabolism (which may be superimposed by bacterial growth). Bacterial growth is most rapid on free surfaces [42, 60] such as the inner walls of the animal chamber, but also in the stirring chamber of the POS, valves, connecting tubings, etc.

Some researchers resort to chemical bactericides to eliminate or inhibit the growth of bacteria in the system [1, 4, 9, 32], whereas others determine the bacterial oxygen consumption in a blank experiment and subtract it from the observed total oxygen uptake ([21, 24, 51], Chap. II.3). However, an inspection of the literature suggests that many workers have ignored the part played by bacterial action in oxygen consumption measurements.

2 Antibiotics

The range of application of antibiotics is very wide, and during the past decades new types have been continuously developed, while at the same time significant progress has been made in clarifying their mechanisms of action [23, 25]. Antibiotics have been used in respirometry in the belief that they exert no influence on the oxygen consumption of the experimental animals [18, 32]. It also should be noted that considerable differences exist in the chemical properties of various antibiotics. Their stability and antimicrobial activity depends to a large extent on the pH and temperature of the medium. Hence the physical and chemical properties of the experimental media largely determine the choice of antibiotic. The properties of some antibiotics have been com-

Bacterial Growth and Antibiotics in Animal Respirometry

piled from the literature in Tables 1 and 2 [12, 20, 23, 26, 29, 41, 46, 54]. Not every antibiotic is suitable for a particular experiment; for instance under anaerobic conditions kanamycin, neomycin and streptomycin are inactive.

The mechanism of action also determines the applicability of a specific antibiotic in respirometry. The following classification is given [41]:

- a) Antibiotics interfering with nucleic acid or protein biosynthesis: actinomycins, adriamycin, amikacin, aminoglycosides, bleomycins, capreomycin, chloramphenicol, clindamycin, fusidic acid, lincomycin, macrolide antibiotics, mitomycins, puromycin, rifamycins, sisomicin, streptonigrin, streptozotocin, thiamphenicol, viomycin.
- b) Antibiotics affecting the function of the cytoplasmatic membrane: amphotericin B, candicidin, gramicidins, nystatin, pimaricin, polymyxins, streptomycin, trichomycin, tyrocidins.
- c) Antibiotics interfering with cell wall biosynthesis: bacitracin, D-cycloserine, cephalosporins, fosfomycin, novobiocin, penicillins, ristocetins, vancomycin.

Due to their unspecific attack on cell metabolism, the first two classes not only damage bacteria, but also disturb the metabolism of the experimental animals. The third class of antibiotics specifically inhibits cell wall synthesis in the bacteria and so hardly influences animal cells.

In selecting an antibiotic it is also important to know the antimicrobial spectra of the antibiotics (Table 2). They can be distinguished as follows [54]:

- a) Wide-spectrum antibiotics: ampicillin, cephalosporins, chloramphenicol, rifampicin, tetracyclines.
- b) Antibiotics preferentially effective against gram-negative bacteria: gentamicin, kanamycins, neomycins, paromomycin, polymyxin B and E, streptomycin.
- c) Antibiotics preferentially effective against gram-positive bacteria: bacitracin, erythromycines, fusidic acid, lincomycin, novobiocin, penicillins, peptolid-antibiotics, rifomycin, ristocetin, vancomycin.
- d) Fungicids: amphotericin B, griseofulvin, nystatin, trichomycin.
- e) Antibiotics inhibiting protozoa: fumagillin, trichomycin.

For use in seawater, antibiotics effective against gram-negative bacteria are to be preferred. Penicillin is inapplicable, since it inhibits mucopeptide synthesis and is thus detrimental to gram-positive bacteria only.

Antibiotics can be classified according to their undesirable toxicity to animals [41]:

- a) Antibiotics with low toxicity: cephalosporins, erythromycin, fusidic acid, lincomycins, penicillins, thiamphenicol.
- b) Toxic antibiotics causing reversible and irreversible damage: aminoglycoside antibiotics (e.g. neomycin), amphotericin B, chloramphenicol, novobiocin, polymyxins, viomycin, paromomycin, tyrothricin, nystatin.

203

¹ Institut für Zoologie, Abteilung Zoophysiologie, Universität Innsbruck, Peter-Mayr-Str. 1A, A-6020 Innsbruck, Austria

Polarographic Oxygen Sensors (ed. by Gnaiger/Forstner) © Springer-Verlag Berlin Heidelberg 1983

Table 1. Properties of some antibiotics: solubility in water, optimal pH range for maximum antimicrobial activity, inactivation rate (%) and inactivating agents in aqueous solutions, and cross-resistance (referring to antibiotic No.)

Soluble 4.5–7.5 7 24–25 30 Stable Acid: 100 5.5 7 24–25 30 Stable Acid: 100 6.5 (6–7.5) 6 5 14 Stable Cood 6.5 (6–7.5) 6 5 14 Stable 220–250 7.5–8.5 7 4 7 24 7 Stable 220–250 7.5–8.5 7 4 7 24 8.1 Stable 220–250 7.5–8.5 7 4 2 Stable 220–250 7.5–8.5 7 4 2 Stable 220–250 7.5–8.5 7 4 2 Stable 237 24 1 Stable Base: 0.55 6.1–6.6 7 37 10 60 Stable Cood 7.5–8.4 7.4–8.0 2.5–9 24 1 Stable 238 37 24 1 Stable Cood 8.5 (6–7.3) 7 24 1 Stable 24 4 4 Stable Cood 8.5 (6–7.3) 7 24 1 Stable Cood 9.5 (6–7.3) 7 24 1 Stable Cood 9.5 (6–7.3) 7 24 1 Stable Cood 9.5 (6–7.3) 7 4 2 Stable Cood 9.5 (6–7.3) 7 24 1 Stable Cood 9.5 (6–7.3) 7 24 1 Stable Cood 9.5 (6–7.3) 7 24 1 Stable Cood 9.5 (6–7.3) 7 25 10 840 10 85 81- and trivalent metallic cations Cood 8.5 (1–6.6 7 37 24 1 55 81- and trivalent metallic cations Cood 8.5 (1–6.6 7 37 24 1 55 81- and trivalent metallic cations Cood 8.5 (1–6.6 7 37 24 1 55 81- and trivalent metallic cations Cood 8.5 (1–6.6 7 37 24 1 55 81- and trivalent metallic cations Cood 9.7 5 8.5 7 24 1 5 81- and trivalent metallic cations Cood 9.7 7 5 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	nti	Antibiotic	Solubility		Inactivation rate	ion rate			Inactivating agents	Cross-
Acid: 100 5.5 (6-7.5) 6 5 14 7 20 Lindo 6.5 (6-7.5) 6 5 14 8 Stable PH > 9; high saline concentrations cause 2 h 100 percipitation; sodium thiosulfate; jons of 2 2 1 2 50 Cood 5.7 (6-7.5) 6 5 5 14 8 Stable PH > 9; high saline concentrations cause 2 1 100 percipitation; sodium thiosulfate; jons of 2 2 1 2 2 50 Cood 5.7 (6-7.3) 7 4 7 Stable PH > 9; high saline concentrations cause 2 2 2 2 2 50 Cood 5.7 (6-7.3) 4 4 Stable PH > 8; light 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2			mg cm ⁻³	Hd	Ηď	J.	days	%	0	resistance
Acid: 100 5.5 7 24 7 20 1000 6.5 (6-7.5) 6 5 14 8 5 able pH > 9; high saline concentrations cause 2 14 2 100 precipitation; sodium thiosulface; jons of 2 21 2 22 2 23 2 24 2 8 2 24 2 8 2 25 2 25 2 27 2 27 2 27 2 29 2 29 2 20 20 2 20 20 20 20 20 20 20 20 20 20 20 20 20	$\vec{\vdash}$	Amphomycin	Soluble	4.5-7.5	7	24-25	30	Stable		
1000 6.5 (6-7.5) 6 5 14 Stable precipitation; sodium thiosulfare; ions of heavy metals Cd³*, Mn²*, Zn²*, H²_0²; Good 2 21 2 30 220-250 7.5-8.5 4 7 Stable precipitation; sodium thiosulfare; ions of heavy metals Cd³*, Mn³*, Zn²*, H²_0²; 200-300 5-7 21 2 11 Stable precipitation; sodium thiosulfare; ions of precipitation;	5.	Ampicillin	Acid: 100	5.5	7	24	7	20		[32]
Good 2 2 11 Stable 220–250 7.5–8.5 4 4 4 Stable 200–300 6.5 (6–7.3) 2.5–4.4 7.4–8.0 2.5–9 24 1 Stable 2.5–4.4 7.4–8.0 2.5–9 24 1 Stable 37 24h 36–39 37 24h 37 24h 36–39 Base: 0.55 6.1–6.6 7 37 10h 65 Bi- and trivalent metallic cations 8.6 Good Good Good Good A 7 5 8.5 A 9 7 8 12h 11–23 A 10h 92 Bi- and trivalent metallic cations 8 37 10h 92 A 15 50 Good A 7 5 6.1 A 7 5 70 Bi- and trivalent metallic cations 8 37 10h 92 A 15 50 Good A 7 5 5 6.1 Bi- and trivalent metallic cations 8 37 10h 92 A 15 50	3.	Bacitracin	1000	6.5 (6–7.5)	9	S	14 2 h	Stable 100	pH \geq 9; high saline concentrations cause precipitation; sodium thiosulfate; ions of heavy metals $C4^{2*}$ Mn. ^{2*} Zn^{2*} . H O	
Good 2 21 2 50 220–250 7.5–8.5 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 5.4ble PH > 8; light 200–300 6.5 (6–7.3) 4 2 Stable 2 11–23 Amount of the string of t	4.	Candicidin			7	4	7	Stable	mary moran od , mm , 5m , 11 ₂ O ₂ ,	
220-250 7.5-8.5 4 4 5table pH > 8; light 200-300 6.5 (6-7.3) 4 4 5table PH > 8; light 200-300 6.5 (6-7.3) 4 2 Stable PH > 8; light 200-300 6.5 (6-7.3) 37 12h 11-23 37 24h 1 Stable PH > 9.5; phenylalamine; cyclic amino-stable 2.5-4.4 7.4-8.0 2.5-9 24 1 Stable PH > 9.5; phenylalamine; cyclic amino-stable Base: 0.55 6.1-6.6 7 37 10h 65 Bi- and trivalent metallic cations 8.6 7 37 24h 95 Pi- and trivalent metallic cations 8.5 7 24h 99.7 Pi- and trivalent metallic cations 8.5 7 24h 99.7 Pi- and trivalent metallic cations 8.5 7 4 7 5 6ood 4 7 5 6ood 4 7 5 6ood 4 7 5	5.	Carbenicillin	Good		7	21	2	50		121
220-250 7.5-8.5 4 4 5table 25 PH > 8; light 200-300 6.5 (6-7.3) 4 2 Stable 37 12 h 12 h 200-300 6.5 (6-7.3) 4 2 Stable 37 12 h 12 h 2.5-4.4 7.4-8.0 2.5-9 24 1 Stable 47-55 pH > 9.5; phenylalamine; cyclic amino-sciol Base: 0.55 6.1-6.6 7 30 60 Stable 5 chloramphenicol chloramphenicol Base: 0.55 6.1-6.6 7 37 10 h 65 Bi- and trivalent metallic cations 8.5 7 37 24 h 95 A 4 7 Good 8.5 4 7 5 7 4 7 5 Good 8.5 4 7 5 7 5 7 4 4 7 5 Good 4 7 5 7 5 4 7 5 Good 4 7 5 4 15 5	.9	Carbomycin			5-7	25	11	Stable		[2]
200–300 6.5 (6–7.3) 4 2 Stable 36–39 2.5–4.4 7.4–8.0 2.5–9 24 1 Stable 36–39 2.5–4.4 7.4–8.0 2.5–9 24 1 Stable acids, metal ions do not deactivate acids, metalions do not deactivate acids, metalions do not deactivate acids, metalions do not deactivate acids, metalic cations acids, metali	7.	Cephaloridin	220-250	7.5-8.5		4 25	4 1-4	Stable Stable	pH > 8; light	[8]
2.5-4.4 7.4-8.0 2.5-9 24 1 Stable pH > 9.5; phenylalanine; cyclic amino-7 30 60 Stable acids; metal ions do not deactivate 7 7 37 10 h 65 Bi- and trivalent metallic cations 8.6 8.3 10 h 92 8.5 4-5 h 50 60d	∞ •	Cephalothin	200-300	6.5 (6–7.3)		4 37 37 37	2 12 h 24 h 48 h	Stable 11–23 36–39 47–55		[7, 18]
Base: 0.55 6.1–6.6 7 37 10 h 65 Bi- and trivalent metallic cations 8.6 8.37 24 h 92 8 37 24 h 99.7 8.5 4 7 5 6.0d	9.	Chloramphenicol	2.5-4.4	7.4-8.0	2.5-9 7 7	24 30 5	1 60 700	Stable Stable Stable	pH > 9.5; phenylalanine; cyclic amino-acids; metal ions do not deactivate chloramphenicol	[40]
Good 4 7 2 25 4	Ö	Chlortetracycline	Base: 0.55 8.6	6.1–6.6	r r 8 8 8 8.		10 h 24 h 10 h 24 h 4-5 h		Bi- and trivalent metallic cations	[30, 39]
		Cloxacillin	Good				· / 4			

	[35]	[11, 24, 29, 32–34]	[39]	[6, 28]	[25]			[15, 22, 25, 31]		[15, 25, 31, 39]		[11, 14, 29, 32–34]
D-alanine; Mycobyctin;	pH > 6; Ca ²⁺		More stable than streptomycin; not deactivated by cysteine	Acids; pH < 6; anaerobiosis does not modify activity; not affected by metallic ions		$p_{ m H} > 7$	$pH < 6; proteins \ and \ an aerobiosis \ do \ not \\ inhibit \ the \ activity$	Na., K., Ca-salts; CO ₃ ²⁻ , SO ₄ ²⁻ , CI ⁻ , PO ₄ ³⁻ , NO ₃ ⁻	Purine	Phosphate, citrate, chloride, Mg^{2+} , anaerobiosis, CO_2^- , K^+ , Ni^{2+} , Fe^{2+} , PO_4^- , Mg^{2+}	Erythromycine	pH 4-5; kanamycin; streptomycin
25 38	Stable 10 57 50	Stable 70 10	Stable Stable	100 14 Stable	Stable	15		Stable	Stable	1.57 1.9 2.8	Stable	20 50 50
7	14 3 16	2 9 14	06	1-3 h 1 56	30	36 h				180 180 120	180	5 2.5
37	30 30 30	24 4 4	4 28	25 -25-+4	25					25 37 45	70	4 25 37
L L	9776		4-7	6 7 8.5	2-9			2.2-10	3-8.8	2.6–7.9 2.6–7.9 2.6–7.9		7-7.2 7-7.2 7-7.2
6.4-7.4	6.5-7.5		7.5-8	8-8.5		9	7.4-8.0	7.8		7.6–8.0	8-8.5	
100	^ \	150	250-500	Base: 2.1 10-20-200	Soluble	Alkali-salt	High	Sulphate: > 20	0.01	Sulphate: 360	Base: 500	Good
12. Cycloserine	13. Colistin	14. Dicloxacillin	 Dihydro- streptomycin 	16. Erythromycin	17. Framycetin	18. Fusidic acid	Gabromycin	20. Gentamycin	21. Griseofulvin	22. Kanamycin	23. Lincomycin	24. Methicillin
12.	13.	14.	15.	16.	17.	18.	19.	20.	21.	22.	23.	24.

Cross- resistance	[14, 15, 32–34, 39]	•		[16]	[11, 14, 24, 32–34]	[10, 40]		[33, 34]	[32, 34]	[32, 33]	[13]	[34]	6
Inactivating agents	Anaerobiosis, RNA, glucose, CI-, Na*, K*, PO ₄ 3-, Fe ^{2,3+} , Mg ²⁻ , Aluminium ions	pH > 9, < 4	Bivalent metallic cations			Bi- and trivalent metallic cations		pH < 4, pH > 8, Pb, Cu, Hg, Zn, Sn, Cd, some amino acids, H_2 O ₂ , rubber, glycerine, ethyl alcohol			pH 2, 8; bivalent cations;		
%	Stable	Stable	50	Stable Stable Stable	38	34 66 25 75		Stable 50	50	Stable	Stable	12	C+010
days	360	Н	09	1 21	15 7 15	10 h 1 10 h 1 10 h		3-7	33 h		360	7	700
ion °C	25	37	24	37 24 25	24 24 24	37 37 37 37		4 24-25	37	4-35	38	24	25
Inactivation pH	∞	7	7-10	2.2–9 5–7 9		~ ~ ∞ ∞		6.5	5	6.5	L-9	9	,
Hq	7.4–8.0	5.5-6.0	5.5	8.0	9.9-9	6.1-6.6	7.6-8.0	6-6.5			6.5-7.5		6.0-7.4
Solubility mg cm ⁻³	6.3	0.21	100	Base: 5 Phosphate: > 1000	Good	Base: 0.6 6.9	Good	> 20	> 750	Good	25	Good	5%
Antibiotic	25. Neomycin B	26. Nitrofurantion	27. Novobiocin	Oleandomycin	29. Oxacillin	30. Oxytetracyclin	31. Paromomycin	Penicillin G	33. Penicillin V	34. Pheneticillin	35. Polymyxin B	36. Propicillin	37. Rifomvcin
Anti	25.	26.	27.	28.	29.	30.	31.	32.	33.	34.	35.	36.	37.

	[15]	[10, 30]	[22, 39]
	pH < 2; pH > 8.5; t > 28°C; KMnO ₄ ; KJO ₄ ; H ₂ O ₂ , HNO ₃ , ascorbic acid, glucose, NO ₃ , OCN ⁻ , Mg, Ca, DNA, urea, anaerobiosis	Bi- and trivalent cations	6 < Hd
50	Stable Stable Stable	2 42 36 82	10 Stable
20	90	10 h 24 h 10 h 24 h	9
30	28 4 15	37 37 37	37
00	4-7 7-7	r r 8 8	3-7
> 8.0	7.5-8.0	6.1-6.6	8.0 7.4–8.4
225	250-500	Base: 1.7 10.9	> 100
38. Spectinomycin	39. Streptomycin	40. Tetracycline	41. Vancomycin 42. Viomycin

Table 2. Antimicrobial spectra of some antibiotics (+ efficient, ± poorly efficient, - inefficient against gram-positive or gram-negative bacteria), minimal inhibitory concentrations and development of resistance (for bacteria)

Antibiotic	В	acteri	a		Fungi	Protozoa	Resistance
	G	+ G-	G ⁺	G-			
			$\mu \mathrm{g~cm^{-3}}$	-	$\mu {\rm g~cm^{-3}}$	$\mu \mathrm{g~cm^{-3}}$	
Adicillin	±	±	5-10	1.5			
Amphomycin	+	-	2.5 - 6.2		Resistant	Resistant	
Amphotericin B	-	-				Nearly all res.	None
Ampicillin	+	±	0.01 - 1.5	> 8		,	121222
Bacitracin	+	-	0.07 - 27	-10000	Resistant	Nearly all res.	Slow
Boromycin	+		0.05		Active		
Candicidin	-	5 ==			0.5 - 50		
Capreomycin	+	-	0.5 - 20		3440-40-2-02-02-0		Rapid
Carbenicillin	+	±	0.2 - 1.56	250			***FIG
Carbomycin	+	200	2-5		Resistant	30-250	
Cephalexin	+	±	50	50			
Cephaloridin	+	±	0.01 - 5	0.1 - 100	Resistant	Resistant	Slow
Cephalothin	+	±	0.02 - 8	>100	resistant	Resistant	Slow
Chloramphenicol	+	+	0.5 - 20	0.4-30	Resistant	125-2000	Slow
Chlortetracycline	+	+	0.3-20	0.1-50	Resistant		Moderate-fa
Cloxacillin	+	_		0.1 50	Resistant	23-1000	Moderate-1a
Colistin	_	+		0.1 - 50	20	125	Slow
Cycloserin	±	+	6.5-50	<125	20	123	Slow
Demethyl-			0.0 00	125			Slow
chlortetracycline	+	+	0.3 - 20	0.1 - 50	Dogistant	Moonles all and	M - 1 t - C
Dicloxacillin	+	-	0.0 20	0.1-30	Resistant	Nearly all res.	Moderate-la
Dihydrostrepto-							
mycin	±	+	1-25	0.5 - 25	Dogistant	David	D
Erythromycin	+		0.01-2	0.5-25	Resistant		Rapid
Framycetin	+	+	0.01-2		Resistant	Nearly all res.	•
Fumagillin	+	_			Danistant	T.CC	Slow
Fusidic acid	+		0.2 - 10		Resistant Resistant		D : 1
Gabromycin	+	+	2-7	1-5	Resistant	Resistant	Rapid
Gentamycin	+	+	0.8-40	1-50	>100	D	None-slow
		•	0.0-40	1-30	>100-	Resistant	Moderate-fa
Gramicidin	+	-	1 - 10		>1000		
Griseofulvin			1-10		0.2.15		
Hamycin		_			0.2-15		None
Kanamycin	±	+	4-500	0.5-15	0.01-4	NT. 1 11	** *
Kitasamycin		_	0.2-3	0.5-15		Nearly all res.	Moderate-fas
incomycin	+		0.2-3 $0.04-32$	> 200	Resistant		
Ayxin	+	+	0.04 - 32 $0.6 - 2.5$	> 200	Resistant		Rapid
Vactins		_	1.0	0.4 - 3.2	1-10		
Veomycin B		+	0.5-150	0.5.15	10		DOMESTICAL TO SEE 1
Vitrofurantoin		+	2-3	0.5-15	Resistant	40-3000	Moderate-fas
		+	0.1-6	0.002-20	10 1000		Slow
Vovobiochi Vystatin		T —	0.1-6	2-25	10-1000		Rapid
, y ocutini	_	_			0.6-5- 1 10-30	250	None
leandomycin	+	_	0.5 - 3		10-30		Danid
	+	-	0.06-1.5				Rapid
		_	0.3-20				

Table 2 (continued)

Bacterial Growth and Antibiotics in Animal Respirometry

Antibiotic	Ba	cteri	a		Fungi	Protozoa	Resistance
	G ⁺	G-	G ⁺ μg cm ⁻³	G ⁻	$\mu {\rm g~cm^{-3}}$	μg cm ⁻³	
Paramomycin	-	+		0.8-25	Resistant	Nearly all res.	Moderate-fast
Penicillin G	+	±	0.006 - 3	>100	Resistant	Resistant	Slow
Penicillin V	+	-	0.01 - 8				Slow
Peptolid-antibiotics	+	5	0.1 - 4		Resistant	Resistant	Moderate-fast
Phenethicillin	+	±	0.03 - 4	>100			
Pimaricin					1 - 12	Nearly all res.	
Polymixin B	_	+		0.05 - 5	125 - 250	125	Slow
Propicillin	+	±	0.03 - 33				
Rifamid	+	±	0.03 - 10	25 - 250			Rapid
Rifampicin	+	±	0.01 - 0.1	1 - 50	>100		Rapid
Rifamycin	+	±	0.03 - 0.1 (10)	25-250			Rapid
Ristocetin	+	1000	0.5 - 4		Resistant	Resistant	None
Rolitetracycline	+	+	0.3 - 20	0.1 - 50	Resistant	Nearly all res.	Moderate fast
Spectinomycin	_	+		4 - 500	Resistant	Resistant	Rapid
Spiramycin	+	_	0.1 - 5				Moderate-fast
Streptomycin	±	_	1 - 25	0.5 - 25	Resistant	Resistant	Rapid
Tetracycline	+	+	0.3 - 20	0.1 - 50	Resistant	60-250	Moderate-fast
Trichomycin					0.6 - 10	250	None
Tyrothricin	+	1000	1 - 10				None-slow
Vancomycin	+	-	0.5 - 3		Resistant	Resistant	None
Viomycin	_	+		$^{2-12}$			Rapid

Table 3 gives examples of the effect of individual antibiotics and their combinations on various aquatic animals. Penicillin has the lowest toxicity, which makes possible its use even in high doses. Kanamycin, neomycin, polymyxins and streptomycin may lead to a neuromuscular block. Kanamycin is less toxic than neomycin and paromomycin, but should not be used in combination with either of these since this leads to a considerable increase in toxicity. A combination of streptomycin and dihydrostreptomycin reduces the toxicity of the individual components. Of the tetracyclines, tetracycline, followed by oxytetracycline, combine optimal antibiotic properties, i.e., low toxicity and a high bactericidal effect. Streptomycin, chloramphenicol, kanamycin, and neomycin are antibiotics with long half-lives in seawater, but due to their high toxicity to invertebrates, their dosage may be critical [12]. Auromycin, terramycin, and polymyxin B are effective against marine bacteria but are toxic to phytoplankton, while penicillin, kanamycin, neomycin, and streptomycin are suitable for phytoplankton cultures [3, 47]. Chlortetracycline, oxytetracycline, bacitracin, carbomycin, Oleandomycin, tetracycline, erythromycin and penicillins have short half-lives in solution, and are therefore unsuitable for long-term experiments [12]. Antibiotics may not only disturb oxygen uptake in animals but also induce anoxic contributions to the total metabolic rate [19].

Table 3. Effect of antibiotics on physiological parameters in animals

Genera	Stage ^a	Antibiotics ^b	μg cm ⁻³	Days ^c	Parameter	Effect ^d	Ref.
Coelenterata							
Tubularia	a	PEN + STR	100 + 100	1	Regeneration	+	[17]
Tubularia	a	PEN + STR	400 + 400	1	Regeneration	+	[17]
Nemertina					₩.		1874 THE
Amphiporus	a	STR + DISTR	750-5000	10	Regeneration	0	[52]
Cerebratulus	a	STR + DISTR	750-5000	10	Regeneration	0	[52]
Lineus	a	STR + DISTR	750-5000	10	Regeneration	0	[52] [52]
Annelida		SCOOLSES STREET	, , , ,	10	Regeneration	U	[32]
Cognettia	a	FRA	250	4	C 11	0.00	
Cognettia	a	PEN	250	4	Growth	+	[28]
Cognettia	a	STR	250	4	Growth	+	[28]
Lumbriculus	a	STR + NEO	20-200	4	Growth	+	[28]
	u.	DIK · NEO	20-200		Heat dissipation	+	[19]
Mollusca	2	2					
Adula	1	PENG + STR	50 + 100	5	Survival	0	[39]
Australorbis	1	PENG + STR	60*+100	3 - 4	Growth	-	[8]
Australorbis Australorbis	1	STR	10	3-4	Growth	0	[8]
Austraiorbis Mya	1	PENG	60*	3-4	Growth	0	[8]
Mya Ostrea	1	PEN + STR	30*+100		Growth	0	[49]
Ostrea	1	PENG	30*	7	Yield of spat	+	[53]
Ostrea	1	PENG	30*	7	Growth	0	[53]
Tritona	1	PENG + STR	30*+50	7	Growth	-	[53]
Tritona	1	STR + PENG	50 + 60	24	Growth	_	[27]
	1	STR + PENG	50 + 60	24	Survival	+	[27]
Crustacea							
Artemia	1 + a	PENG	150	37	Growth	0	[12]
Artemia	1 + a	PENG	200	7	Growth		[12]
Artemia	1	STR	200	37	Growth	0	[12]
Artemia	1 + a	STR	300	7	Growth	-	[12]
Artemia	a	STR	250	37	Growth	0	[12]
Artemia	1	POL	30	37	Growth	0	[12]
Artemia	1 + a	POL	50	7	Growth	1-1	[12]
Artemia	a	POL	40	37	Growth	0	[12]
Artemia	1 + a	CHL	25	37	Growth	0	[12]
	1 + a	CHL	50	7	Growth	=	[12]
***********	1 + a	NAR	10	37	Growth	0	[12]
	1	NAR	15	7	Growth	_	[12]
Artemia	a	NAR	20	7	Growth	_	[12]
* a filtra and the	l + a	CAN	2.5	37	Growth	0	[12]
	1	CAN	3	7	Growth	-	[12]
	a	CAN	5	7	Growth	-	[12]
	1 + a	NYS	10	37	Growth	0	[12]
# 10 YOUR PROPERTY AND ADDRESS OF THE PROPERTY ADDRESS OF	1	NYS	15	7	Growth	-	[12]
	a	NYS	30	7	Growth	25	[12]
•	1 + a	KAN	250	37	Growth	0	[12]
		KAN	500	7	Growth	= 8	[12]
100		KAN	750	7	Growth	(- :	[12]
		NEO	5	37	Growth	0	[12]
December of the second		NEO	10	7	Growth	-	[12]
ir contiu	a	NEO	15	7	Growth	_	[12]

Table 3 (continued)

Genera	Stage ^a	Antibiotics ^b	$\mu {\rm g~cm^{-3}}$	Days ^c	Parameter	Effect ^d	Ref.
Artemia	1 + a	NAL	50	37	Growth	0	[12]
Artemia	1	NAL	100	7	Growth	-	[12]
Artemia	a	NAL	150	7	Growth	_	[12]
Artemia	1 + a	TET	5	37	Growth	0	[12]
Artemia	1	TET	7.5	7	Growth	-	[12]
Artemia	a	TET	9.5	7	Growth	223	[12]
Artemia	1 + a	TRI	25	37	Growth	0	[12]
Artemia	1	TRI	50	7	Growth	_	[12]
Artemia	a	TRI	75	7	Growth	: :	[12]
Artemia	1 + a	MET	500	37	Growth	0	[12]
Artemia	1 + a	MET	1000	7	Growth	: = :	[12]
Calanus	a	CHL + STR	50 + 50		Respiration	0	[32]
Calanus	a	PEN	50	1	Feeding	+	[32]
Calanus	a	CHL	50	1	Feeding	;—;	[32]
Calanus	a	CHL + STR	50 + 50	1	Feeding	-	[32]
Calanus	a	STR	50	1	Feeding	0	[32]
Cancer	1	STR + PEN	100 + 100	40	Survival	+	[15]
Cancer	1	STR + PEN	50 + 50	40	Survival	+	[15]
Cancer	1	STR + PEN	10 + 10	40	Survival	+	[15]
Cancer	1	STR + PEN	100 + 100	100	Survival	+	[16]
Cancer	1	CHL	1	38	Survival	+	[16]
Cancer	1	CHL	5	38	Survival	+	[16]
Cancer	1	CHL	10	38	Survival	+	[16]
Cancer	1	STR + PEN	100 + 100	38	Survival	+	[16]
Cancer	1	CHL + KAN	1 + 100	30	Survival	-	[16]
Cancer	1	KAN	100	30	Survival	_	[16]
Cancer	1	CHL + NEO	1 + 50	27	Survival	7-0	[16]
Cancer	1	NEO	50	27	Survival	_	[16]
Cancer	1	PEN + STR	8 + 130		Survival	+	[39]
Daphnia	a	DISTR + CHL	25 + 25	1	Respiration	0	[59]
Daphnia	a	DISTR + TET	25 + 25	1	Respiration	0	[59]
Daphnia	a	DISTR + TET	50 + 50	1	Respiration		[59]
Eurytemora	a	CHL	20	-	Respiration	0	[22]
Palaemon	1	PENG + STR	30*+ 50	30-36	Survival	0	[57]
Pleuroncodes	1	PPEN + PEN	75 + 25	74	Growth	_	[5]
Temora	a	CHL	50	25.775	Growth	220	[22]

c Days of treatment

a Abbreviations: a = adult; 1 = larval Abbreviations: BAC = bacitracin; CAN = candicidin; CHL = chloramphenicol; DISTR = dihydrostreptomycin; FRA = framycetin; KAN = kanamycin; MET = methenamine-mandalate; NAL = nalidixin; NAR = naramycin; NEO = neomycin; NYS = nystatin; PEN = penicillin; PENG = penicillin G; POL = polymixin; PPEN = procaine penicillin; STR = streptomycin; TET = tetracyclin; TRI = trichomycin

d The effect is related to the parameter against the untreated control (+ positive, - negative, 0 no effect)

^{*} Calculated from international units

3 Other Chemotherapeutics and Bactericidal Agents

Before the discovery of antibiotics chemotherapeutics were of considerable importance. However, their use in respirometry is not recommended due to their high toxicity.

Chlorine is unsuitable in respirometry and its toxicity is dependent on temperature and species [34, 44]. Significant reductions in standard respiration rates were measured in larval lobsters [7]. Increasing concentrations of residual chlorine inhibit larval development and cause morphologic deformations in some fish [36]. These effects have to be borne in mind when using chlorinated tap water, in which chlorine is more persistent than in seawater [14].

Ozone is a potent germicide [20, 37, 55]. Its efficiency is affected by temperature and pH, and it is unstable in water where the contact time is more critical than the quantity of ozone used [48]. Its toxicity is still under debate [13, 50, 56]; but due to the liberation of oxygen it interferes with the polarographic measurement.

UV-irradiation is an effective disinfectant [6], but since its penetration is limited, bacteria which adhere to the inner surfaces of the respirometer are barely affected.

4 Bacterial Growth and Respiration

Inoculation of the respirometer with bacteria occurs via the medium and the experimental animals. Bacteria are seldom freely suspended in natural waters but are generally attached to solid suspended particles [11, 35, 45]. They can be eliminated by filtering the test water $(0.45~\mu m)$. Bacteria growing on the body surfaces of the test animals, however, cannot be eliminated. Rinsing the animals in a bactericide solution is not advisable, because handling and the toxicity of the solution are likely to induce stress in the animals. Bacterial respiratory inhibition by antibiotics is dependent on the growth phase and the position within a phase of growth. Therefore uniform suppression of bacterial respiration cannot be assumed [58]. If the bacteria are inhibited, fungal growth increases [12, 38]. On the other hand, antibiotics may interfere with the metabolism of the experimental animal. In some cases the total metabolism of the animal in the experiment may be considerably increased, as in Lumbriculus variegatus and Salvelinus fontinalis ([19]; Chap. II.3).

The estimation of microbial respiration in a control chamber without experimental animals is not to be recommended, because the conditions for the growth of bacteria are not identical in test and control (bacteria inoculation by experimental animals, availability of soluble excreta). The subtraction of the mean background O₂-consumption rate, i.e. (initial + final background): 2, from the oxygen uptake rate is unsuitable if the respirometer has been previously sterilized or cleaned, since this does not take into consideration the form of the bacterial growth curve during the experiment (Figs. 1 and 2).

5 Experimental Observations

A possible solution is suggested by considering the growth curve of bacteria. After an initial log-phase and following the exponential log-phase bacterial growth in a chemostats culture remains constant (stationary phase). This microbiological principle proved to be valid for bacterial growth in a respirometer.

Oxygen consumption measurements were made with an automated polarographic respirometer (Chap. II.1). This is an intermittently closed system in which any of the four chambers can be automatically closed in sequence, while flow is maintained through the other three chambers. The oxygen uptake by eight prawns (*Palaemonetes antennarius*) was recorded as the decrease in dissolved oxygen during "closed" periods of 12 min. The interval between two closed periods of the same chamber was 1 h. The prawns were repeatedly put into the respirometer for 1 h and subsequently removed for 1 h to permit measurement of the bacterial respiration in the respirometer. In this way the bacterial contamination by the experimental animals and the presence of the excreta of the prawns were guaranteed. Animals were put into the compartments of

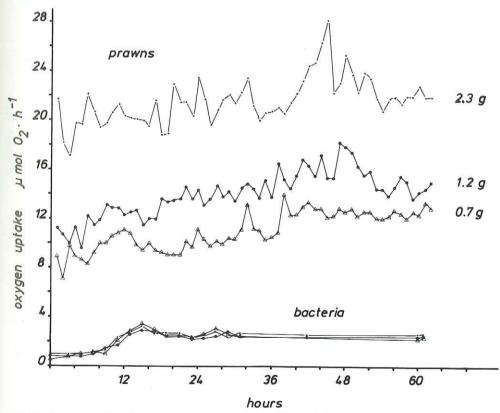


Fig. 1. Oxygen uptake of prawns uncorrected for bacterial metabolic rate and the respective simultaneous oxygen consumption rate of the bacteria in the system. The total biomass (wet weight) is given for each size group

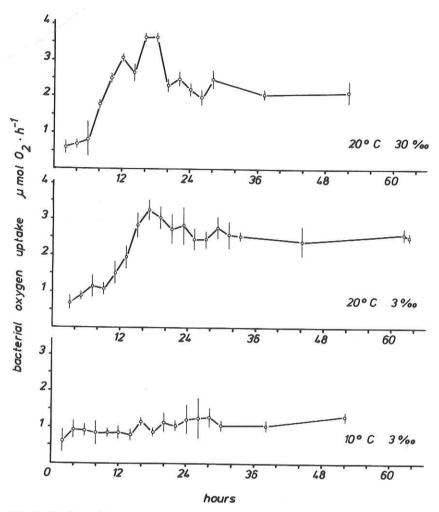


Fig. 2. The bacterial growth curve in the respirometer discontinuously inoculated with experimental animals during different conditions of salinity and temperature. Means ± S.D. are given

cages made of stainless steel netting to permit their rapid removal from the respirometer.

The bacterial growth in the respirometer increases exponentially after a lag-phase of 5 to 6 h, overshoots, and stabilizes after 12 to 20 h irrespective of animal size (Fig. 1). Hence, true animal respiration rates are obtained by determining the final bacterial background rate and substracting it from the total oxygen uptake in the experiment discharging the initial period. To estimate the initial rates, the changing bacterial respiration rate should be subtracted from the total oxygen consumption rate.

The curve of bacterial growth in the respirometer should be determined under experimental conditions. Salinity in contrast to temperature affects bacterial growth only slightly (Fig. 2). The microbial growth curve must also be determined for each

system and respirometer because the relative surface area and the materials of the respirometer exert a considerable influence [31, 33, 43]. The results also confirm that over the first 2-5 h no significant bacterial growth is observable [18] although in long-term measurements bacterial respiration can attain high levels. The bacterial oxygen consumption may be as high as 30% of that of the specimens under observation (Fig. 1). The population densities of bacteria occupying the surfaces are relatively constant and independent of the quantity of the available energy source [40, 48].

The bacterial respiratory effect is decreased by reducing the surfaces in the respirometer, avoiding plastic materials in the construction, maintaining a flow-through and increasing the biomass in the animal chamber.

6 Bacterial Interference with Stability of POS

A new membrane mounted on the POS constitutes a surface which may be gradually overgrown by bacteria. There was a correlation between deviation of the POS signal from the original calibration point and bacterial oxygen uptake (Fig. 3). This may be

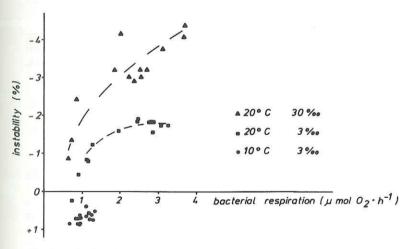


Fig. 3. Deviation of the POS stability from the beginning of the experiments as a function of bacterial respiration. Without intermittent calibrations the expected concentration of oxygen, c_p' at time t would be

$$c'_t = I_t \times \frac{c_0}{I_0}$$
,

where $c_{\mathbf{Q}}$ and $I_{\mathbf{Q}}$ are the oxygen concentration and signal of the POS at the calibration point. The relative instability of the POS is then calculated as

$$\frac{c_t'-c_t}{c_t} \times 100,$$

The broken lines were fitted by eye to indicate the trends considered significant

due (1) to O_2 -depletion of the membrane boundary layer by bacterial respiration, and (2) to increased thickness of the diffusion layer (membrane + bacteria) which reduces the membrane permeability. This source of error should be considered in long term in situ measurements (Part III) where uncontrolled bacterial Aufwuchs on the membrane will occur.

7 Conclusions

The bacterial problem in the respirometry of aquatic animals arises from excessive bacterial growth on the inner walls of the respirometer, especially in long-term measurements. The use of antibiotics may be considered in combatting this problem, but is limited by their physicochemical properties and because they are often more toxic to the animals than to the bacteria. Chemotherapeutics are also unsuitable in respirometry because of their high toxicity.

The time course of bacterial growth shows that the bacterial oxygen consumption rises to constant levels within 12 to 20 h and has to be subtracted from the total oxygen consumption rate. By this method no chemicals need be used to inhibit bacterial growth, thus avoiding any possible influence on the experimental animals.

Acknowledgments. I wish to express my gratitude to E. Gnaiger for proposing critical comments and suggestions on the manuscript. The study was supported by the Oesterreichische Nationalbank, project no. 1208 and by the Österreichischer Forschungsfond, project no. 3307.

References

- Alcaraz M (1974) Respiración en crustáceos: Influencia de la concentración de oxígeno en el medio. Invest Pesq 38:397-411
- Bell GR, Hoskins GE, Hodgkiss W (1971) Aspects of the characterization, identification, and ecology of the bacterial flora associated with the surface of stream-incubating Pacific Salmon (Oncorhynchus) eggs. J Fish Res Board Can 28:1511-1525
- Berland BR, Maestrini SY (1969) Study of bacteria associated with marine algae in culture.
 II. Action of antibiotic substances. Mar Biol 3:334-335
- 4.Booth CE, Mangum CP (1978) Oxygen uptake and transport in the lamellibranch mollusc Modiolus demissus. Physiol Zool 51:17-32
- Boyd CM, Johnson MW (1963) Variation in the larval stages of a decapod crustacean, Pleuroncodes planipes Stimpson (Galatheidae). Biol Bull 124:141-152
- Brown C, Russo DJ (1979) Ultraviolet light disinfection of shellfish batchery sea water. I. Elimination of five pathogenic bacteria. Aquaculture 17:17-23
- Capuzzo JM (1977) The effects of free chlorine and chloramine on growth and respiration rates of larval lobsters (Homarus americanus). Water Res 11:1021-1024
- Chernin E (1959) Cultivation of the snail Australorbis glabratus under axenic conditions. Ann NY Acad Sci 77:237–245
- Childress JJ (1975) The respiratory rates of midwater crustaceans as a function of depth of occurrence and relation to the oxygen minimum layer of Southern California. Comp Biochem Physiol 50A:787-799

- Colberg PJ, Lingg AJ (1978) Effect of ozonation on microbial fish pathogens, ammonia, nitrate, nitrite and BOD in simulated reuse hatchery water. J Fish Res Board Can 35:1290– 1296
- 11. Cooke WB (1956) Colonization of artificial bare areas by microorganisms. Bot Rev 22:613-638
- 12. D'Agostino A (1975) Antibiotics in culture of invertebrates. In: Smith WL, Chanley MH (eds) Culture of marine invertebrate animals. Plenum Press, New York London, pp 109-133
- Dungworth DL, Cross CE, Gillespie JR, Plopper CG (1975) The effects of ozone on animals.
 In: Murphy JS, Orr JR (eds) Ozone chemistry and technology. Franklin Inst Press, Philadelphia, pp 29-54
- Duursma EK, Parsi P (1976) Persistence of total and combined chlorine in sea water. Neth J Sea Res 10:192-214
- 15. Fisher WS, Nelson RT (1977) Therapeutic treatment for epibiotic fouling on Dungeness crab (Cancer magister) larvae reared in the laboratory. J Fish Res Board Can 34:432-436
- Fisher WS, Nelson RT (1978) Application of antibiotics in the cultivation of Dungeness crab, Cancer magister. J Fish Res Board Can 35:1343-1349
- 17. Fulton C (1959) Re-examination of an inhibitor of regeneration in *Tubularia*. Biol Bull 116: 232-238
- 18. Giese AC, Farmanfarmaian A, Hilden S, Doezema P (1966) Respiration during the reproductive cycle in the sea urchin Strongylocentrotus purpuratus. Biol Bull 130:192-201
- 19. Gnaiger E (1980) Energetics of invertebrate anoxibiosis: Direct calorimetry in aquatic Oligochaetes. FEBS Lett 112(2):239-242; Pharmacological application of animal calorimetry. Thermochim Acta 49:75-85
- Goldberg HS (1959) Antibiotics, their chemistry and nonmedical uses. Van Nostrand Co, New Jersey
- Green JD, Chapman MA (1977) Temperature effects on oxygen consumption by the copepod Boeckella dilatata. N Z J Mar Freshwater Res 11:375-382
- 22. Gyllenberg G (1973) Comparison of the Cartesian diver technique and the polarographic method, an open system, for measuring the respiratory rates in three marine copepodes. Commentat Biol 60:1-13
- Hahn FE (1979) Antibiotics, vol V. Mechanism of action of antibacterial agents. Springer, Berlin Heidelberg New York
- 24. Harding GCH (1977) Surface area of the Euphansiid *Thysanöessa raschii* and its relation to body length, weight, and respiration. J Fish Res Board Can 34:225-231
- 25. Hash JH (1972) Antibiotic mechanisms. Annu Rev Pharmacol 12:35-56
- 26. Helwig H (1973) Antibiotika-Chemotherapeutika. Thieme, Stuttgart
- 27. Kempf SC, Dennis Willows AO (1977) Laboratory culture of the nudibranch Tritonia diomedea Bergh (Tritoniidae: Opisthobranchia) and some aspects of its behavioral development. J Exp Mar Biol Ecol 30:261-276
- Latter PM (1977) Axenic cultivation of an enchytraeid worm, Cognettia sphagnetorum. Oecologia 31:251-254
- Lorian V (1966) Antibiotics and chemotherapeutic agents in clinical and laboratory practice.
 CC Thomas, Springfield, Illinois, USA
- 30. Lough RG, Gonor JJ (1973) A response-surface approach to the combined effects of temperature and salinity on the larval development of Adula californiensis (Pelecypoda: Mytilidae).
 I. Survival and growth of three and fifteen-day old larvae. Mar Biol 22:241-250
- 31. Marshall KC (1972) Mechanism of adhesion of marine bacteria to surfaces. Gaitherburg, Maryland, USA, Proc 3rd Int Congr Mar Corrosion Fouling, October 1972, pp 625-632
- Marshall SM, Orr AP (1958) Some uses of antibiotics in physiological experiments in sea water.
 J Mar Res 17:341-346
- 33. Meadows PS (1964) Experiments on substrate selection by *Corophium* species: Films and bacteria on sand particles. J Exp Biol 41:499-511
- 34. Middaugh DP, Crane AM, Couch JA (1977) Toxicity of chlorine to juvenile spot, *Leiostomus xanthurus*. Water Res 11:1089-1096
- 35. Moebus K (1972) Factors affecting survival of test bacteria in sea water: marine bacteria, test bacteria and solid surfaces. Helgol Wiss Meeresunters 23:271-285

- 36. Morgan RP, Prince RD (1978) Chlorine effects on larval development of striped bass (Morone saxatilis), white perch (M. americana) and blueback herring (Alosa aestivalis). Trans Am Fish
- 37. Murphy JS, Orr JR (1975) Ozone chemistry and technolog: A review of the literature 1961-1974. Franklin Inst Press, Philadelphia
- 38. Nilson EH, Fisher WS, Shleser RA (1976) A new mycosis of larval lobster (Homarus americanus). J Invertebr Pathol 27:177-183
- 39. Poole RL (1966) A description of laboratory-reared zoeae of Cancer magister Dana, and megalopae taken under natural conditions (Decapoda Brachyura). Crustaceana 11:83-97
- 40. Quastel JH, Scholefield PG (1951) Biochemistry of nitrification in soil. Bacteriol Rev 15:1-23
- 41. Reiner R (1977) Antibiotics. In: Korte F, Goto M (eds) Natural compounds. Part 2: Antibiotics, vitamins and hormones. Thieme, Stuttgart, pp 1-68
- 42. Relini G (1974) Colonization patterns of hard marine substrata. Mem Biol Mar Oceanogr 4
- 43. Sechler GE, Gundersen K (1972) Role of surface chemical composition on the microbial contribution to primary films. Gaithersburg, Maryland, Proc 3rd Int Congr Mar Corrosion Fouling,
- 44. Seegert GL, Brooks AS (1978) The effects of intermittent chlorination on Coho salmon, Alewife, Spottail Shiner and Rainbow Smelt. Trans Am Fish Soc 107:346-353
- 45. Sheldon RW, Evelyn TPT, Parson TR (1967) On the occurrence and formation of small particles in sea water. Limnol Oceanogr 12:367-375
- 46. Spector WS (1957) Handbook of toxicology, vol II. Antibiotics. W.B. Saunders, Philadelphia
- 47. Spencer CP (1952) On the use of antibiotics for isolating bacteria-free cultures of marine phytoplankton organisms. J Mar Biol Assoc UK 31:97-106
- 48. Spotte S (1979) Fish and invertebrate culture. Wiley, New York
- 49. Stickney AP (1964) Salinity, temperature, and food requirements of soft-shell clam larvae in laboratory culture. Ecology 45:283-291
- 50. Stokinger HE (1965) Ozone Toxicology: A review of research and industrial experience, 1954-1964. Arch Environ Health 10:719-731
- 51. Swiss JJ, Johnson MG (1976) Energy dynamics of two benthic crustaceans in relation to diet. J Fish Res Board Can 33:2544-2550
- 52. Tuker M (1959) Inhibitory control of regeneration in nemertean worms. J Morphol 105:569-
- 53. Walne PR (1958) The importance of bacteria in laboratory experiments on rearing the larvae of Ostrea edulis (L.). J Mar Biol Assoc UK 37:415-425
- 54. Walter AM, Heilmeyer L (1969) Antibiotika-Fibel. Antibiotika und Chemotherapie. Thieme,
- 55. Wedemeyer GA, Nelson NC (1977) Survival of two bacterial fish pathogens (Aeromonas salmonicida and the enteric redmouth bacterium) in ozonated, chlorinated, and untreated waters. J Fish Res Board Can 34:429-432
- 56. Wedemeyer GA, Nelson NC, Yasutake WT (1979) Physiological and biochemical aspects of ozone toxicity to Rainbow trout (Salmo gairdneri). J Fish Res Board Can 36:605-614
- 57. Wickins JF (1972) Developments in the laboratory culture of the common prawn, Palaemon serratus Pennant. Fish Invest London Ser II 27(4):1-24
- 58. Yetka JE, Wiebe WJ (1974) Ecological application of antibiotics as respiratory inhibitors of bacterial populations. Appl Microbiol 28:1033-1039
- 59. Zeiss FR (1963) Effects of population densities on zooplankton respiration rates. Limnol
- 60. Zobell CE, Anderson DQu (1936) Observations on the multiplication of bacteria in different volumes of stored sea water and the influence of oxygen tension and solid surfaces. Biol Bull 71:324-342